

Identifying Acids and Bases

Objectives:

1. To use a variety of tests to identify if a **solution** is acidic or basic
2. To use the same tests to identify a series of **House Hold Products** as either acidic or basic.

Materials:

- A. pH paper
- B. phenol-phthalein
- C. methyl orange
- D. blue litmus
- E. red litmus
- F. Magnesium strips

Solutions:

- 1. NaOH solution
- 2. HCL solution
- 3. NH₃ solution
- 4. Ca(OH)₂ solution
- 5. H₂SO₄ solution
- 6. KOH solution

Common Household Products:

- 1. Vinegar
- 2. Easy off
- 3. Ammonia
- 4. Pop
- 5. Lime juice
- 6. Milk of Magnesia

Procedure:

Part 1 – Identify known solutions as acidic or basic

1. Collect 6 dropper plates and using a pencil label 5 wells A to F
2. Place the appropriate tester into the designated well (A=pH paper, B=1 drop phenolphthalein, C=1 drop methyl orange, D=blue litmus, E=red litmus, F=Magnesium strip)
3. For each of the 6 trays place 3 drops of each solution into the wells with the testers
4. Observe and record the results in the data table

Part 2 –Identifying Common House Hold Products as acidic or basic

5. Collect 6 dropper plates and using a pencil label 5 wells A to F
6. Place the appropriate tester into the designated well (A=pH paper, B=1 drop phenolphthalein, C=1 drop methyl orange, D=blue litmus, E=red litmus, F=Magnesium strip)
7. For each of the 6 trays place 3 drops of each House Hold Product into the wells with the testers
8. Observe and record the results in the data table

Data and Observations:

Part 1: KNOWN ACIDS or BASES

Solution	pH value (paper)	Colour in Phenolphthalein	Colour in Methyl Orange	Colour of Blue Litmus	Colour of Red Litmus	React with Mg?	Acid or Base?
NaOH							
HCl							
NH ₃							
Ca(OH) ₂							
HF							
KOH							

Part II: HOUSEHOLD PRODUCTS

Solution	pH value (paper)	Colour in Phenolphthalein	Colour in Methyl Orange	Colour of Blue Litmus	Colour of Red Litmus	React with Mg?	Acid or Base?
Vinegar							
Easy Off							
Ammonia							
Pop							
Lime Juice							
Milk of Magnesia							