Name:		
Blk:	_Date:_	

# Science 8 Notes on Elements

### **Chemical Symbols**

Because elements have different names in different languages, chemists use **international symbols** for them Chemical symbols consist of **one, two or three** letters. The first of which is always **Capitalized** while the second and or third is always **lower case** Ancient names are used as the source of many of the symbols.

Example: Mercury - **Hg -** H**ydra**g**yrum** (*Latin for liquid silver*)

Look at pg 44 Table 2.1 Thirty-five Common Elements By the end of this unit you will know all of these

#### **Some Common Elements**

Hydrogen (H)- a non-metal (gas)

Colourless, odourless, tasteless, and highly flammable Makes up over **90 percent** of the atoms in the universe Lighter than air

#### **Iron (Fe)** – a metal

Is mixed with carbon to make <u>steel</u> Good structural material, but can <u>rust</u> when mixed with water or oxygen

#### Oxygen (O) - a non-metal (gas)

Necessary for our existence (required for the process of <u>respiration</u>)
Produced by plants in a process known as <u>photosynthesis</u>

#### **Other Common Elements**

Sodium (Na) - soft metal that reacts with water
Chlorine (Cl) - yellow-green gas that is highly toxic
Mercury (Hg) - liquid at room temperature metal.
Silver (Ag) - precious metal mined in British Columbia
Silicon (Si) - brittle, grey, semiconductor that is second most common element in Earth's crust.

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