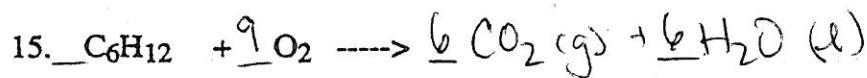
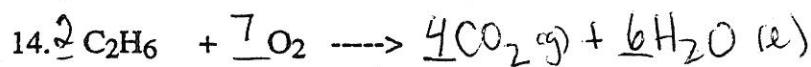
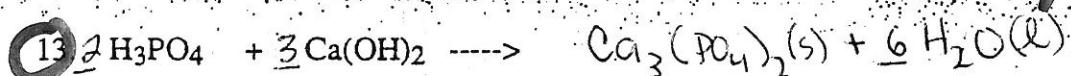
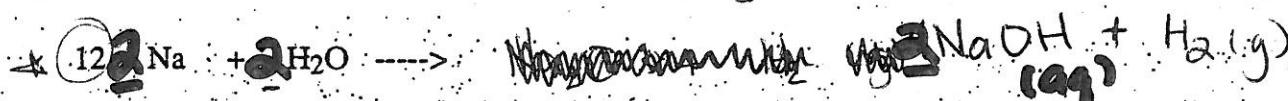
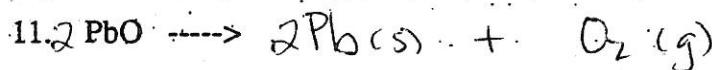
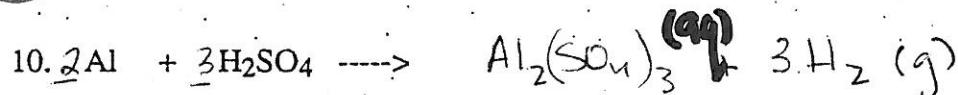
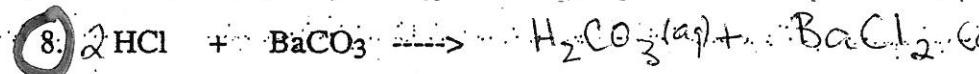
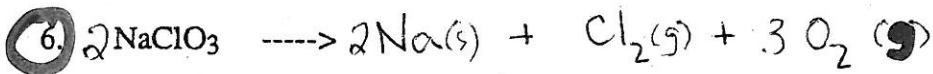
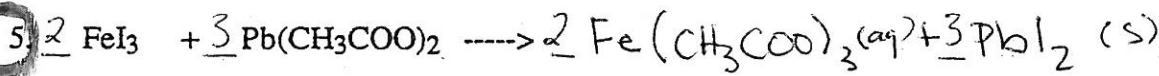
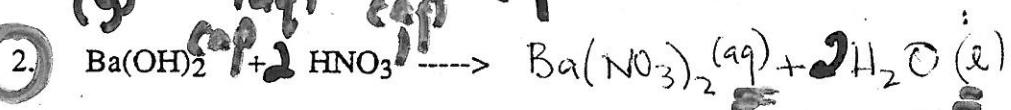
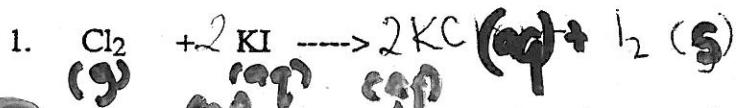


Complete and balance the following equations:



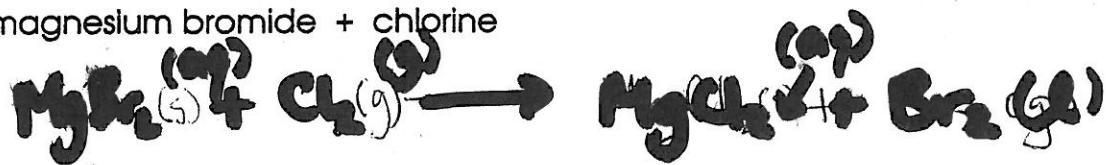
1

REDICTING PRODUCTS OF CHEMICAL REACTIONS

Name Key

Predict the products of the reactions below. Then, write the balanced equation and classify the reaction.

1. magnesium bromide + chlorine



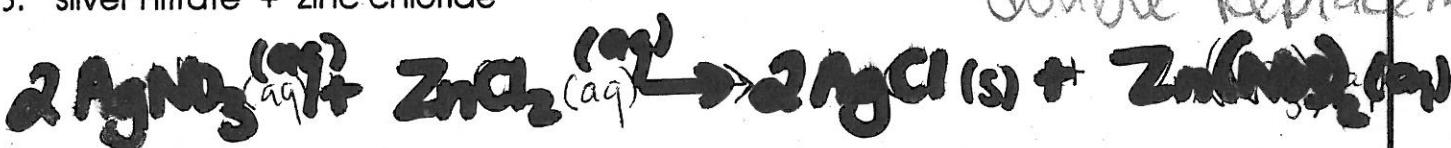
Single Replacement

2. aluminum + iron (III) oxide



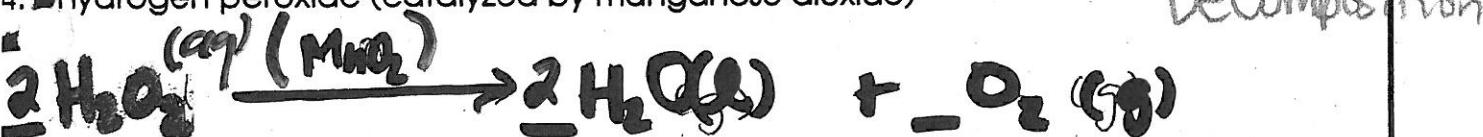
Single Replacement

3. silver nitrate + zinc chloride



double Replacement

4. hydrogen peroxide (catalyzed by manganese dioxide)



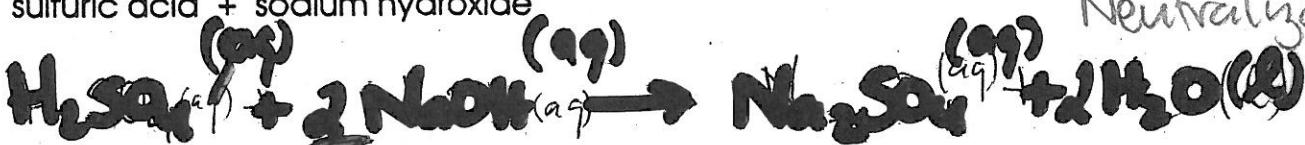
Decomposition

5. zinc + hydrochloric acid



Single Replacement

6. sulfuric acid + sodium hydroxide



Neutralization

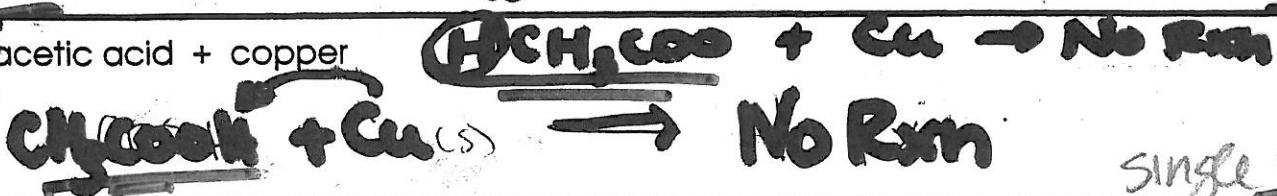
7. sodium + hydrogen



Synthesis

13

8. acetic acid + copper



No Rxn

single replacement