Name:		
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## CHEMISTRY 11 Organic Chemistry Lesson #5

The 8 Functional Groups Continued.
In the previous lesson we discussed how to name an alcohol. In this lesson we will learn will learn how to name and
Rules for naming an ORGANIC ACID: Recall that an organic acid is a hydrocarbon that ends in a -COOH group.  1. 2.
Example 1: Name the following organic acids: CH <sub>3</sub> -CH <sub>2</sub> -COOH CH <sub>3</sub> -COOH
Rules for naming an ESTER: Recall that an ESTER is a hydrocarbon with a -COO in backbone  1.  2.
Example 2: Name the following esters:
CH <sub>3</sub> -COO- CH <sub>2</sub> -CH <sub>3</sub> CH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> -COO- CH <sub>2</sub> -CH <sub>3</sub>
ESTER FORMATION:  Combining an( )with an( )  in the presence of an INORGANIC ACID such asor  forms an ESTER and WATER

Example 3:

**Example 4:** Show the skeletal equation for the formation of the following esters in the presence of hydrochloric acid. Include the name for the alcohol and organic acid.

a. Ethyl propanoate

b. butyl butanoate

c. methyl octanoate

**HOMEWORK**: Exercises 34-36