Name:_____ Blk: Date:_____

Chemistry 11 Organic Chemistry Lesson #3 NAMING MULTIPLE BOND STRUCTURES

Members of the HALOGEN FAMILY can also be added to organic compounds. Because Halogens have a combining capacity of <u>One</u> they form a single bond to attach itself to the compoud.

re fluoro, chloro, bromo, iodo RULES for naming attached halogens: u Change the "ine" to "o" ie flooro, chloro, bromo, todo
I change the "ine" to "o" ie flooro, chloro, bromo, todo
identify location of halogen with a number
when multiple of the same are present use latin petix
List all attached groups alphabetically the FIRST
give the LOWEST possible ADDRESS to the FIRST
group. I-bromo-3-etyl portare 1-bromo-3-styl portare Example 1: Name the following L 3 4 5 $\dot{C}H_3 - \ddot{C}H_2 - \dot{C}H_2 - \ddot{C}H_2 - \ddot{C}H_3$ $\dot{C}H_2 - \ddot{C}H_2 - \ddot{C}H_2 - \ddot{C}H_2 - \dot{C}H_3$ CH₃-CI chloro methane) 2 - fluoro pentane Br $CH_3 - CH_3$ Do EX:17+18 Pg 225 brons ethic RULES FOR NAMING ORGANIC COMPOUNDS WITH MULTIPLE BONDS ANE" is "ENE" present " 1. If at least one double bond is 2. if at least one triple band is present "ANE" is "YNE" 3. Give the LOWEST POSSIBLE ADDRESS to the BOND 4. identify the location of the multiple band mane! by placing a number infront of the parent name! Example 2: Name the following: $CH = C - CH_2 - CH_2 - CH_2 - CH_2 - CH_3$ 2 2 બ $CH_3 - CH = CH_2 - CH_2 - CH_3$ 2- hexene 1-heptyne 23 4 CH=C-CH2-CH3 CH₃ 2- methyl- 1- butene HOMEWORK: Exercises 17 + 18, 22, 23 + 24 a, b, d+e page 228 page 225