Name:
Blk: $\qquad$ Date: $\qquad$
Chemistry 11
Organic Chemistry
Lesson \#2 NAMING BRANCHED HYDROCARBONS
In order to name a single chained hydrocarbon you have to understand that there can be $\qquad$ made to any carbon in an organic molecule,

It is these $\qquad$ that change the name of the structure. The first type of attached group that we are going to learn are called $\qquad$ ,

An ALKYL is simply an ALKANE that has one less $\qquad$ atom, This missing hydrogen frees up a $\qquad$ and allows it to attach itself to the organic molecule,

When naming an ALKYL, change the "ANE" ending of the alkane to "YL"
Fill in the following table with the appropriate expanded structures

| NAME | \# of C | FORMULA |
| :--- | :--- | :--- |
| METHYL |  |  |
| ETHYL |  |  |
| PROPYL |  |  |
| BUTYL |  |  |
| PENTYL |  |  |
| HEXYL |  |  |
| HEPTYL |  |  |
| OCTYL |  |  |
| NONYL |  |  |
| DECYL |  |  |

## RULES FOR NAIMING BRANCHED HYDROCARBONS:

1. 
2. 
3. 
4. 
5. 

IMPT: the atoms in the parent hydrocarbon are numbered to give the LOWEST POSSIBLE NUMBERS to the attached ALKYLS.

Example 1: Name the following branched hydrocarbons


MORE RULES: When more than one attached alkyl is present 1.
2.
3.

Example 2:
$\underset{\mathrm{CH}_{3}}{\mathrm{CH}_{3}-\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}}$

ANOTHER RULE: if the SAME alkyl group is repeatedly attached:
1.
2.
3.

## Example 3.:



HOMEWORK: QUESTIONS 8 and 9 page 221
(b)

(d)

(f)

(h)

(j)

(I)

(f) 5-butyl-6,6-diethyl-3,3,7-trimethyldecane
(g) dimethylpropane (why were no numbers used?)
(h) 4-ethyl-2-methyloctane
(i) hexamethylpentane
(j) 3,6-diethyl-4-methyl-5-propyloctane

