## METRICS AND MEASUREMENT

Name \_\_

the chemistry classroom and lab, the metric system of measurement is used, so it is inportant to be able to convert from one unit to another.

mega	kilo	hecto	deca (da) 10		
(M)	(k)	(h)			
1,000,000	1000	100			
106	10³	10 <sup>2</sup>	10¹		

deci	centi	milli	micro (μ) .000001	
(d)	(c)	(m)		
.1	.01	.001		
10-1	10-2	10-3	10-6	

## **Factor Label Method**

- 1. Write the given number and unit.
- 2. Set up a conversion factor (fraction used to convert one unit to another).
  - a. Place the given unit as denominator of conversion factor.
  - b. Place desired unit as numerator.
  - c. Place a "1" in front of the larger unit.
  - d. Determine the number of smaller units needed to make "1" of the larger unit.
- 3. Cancel units. Solve the problem.

		4						
<b>Example</b> 1: 55 mm = m			<b>Example 2:</b> 88 km = m					
55 merr	1 m	= 0.055 m		88 km	1000 m	= 88,000 m	aut.	
	1000 mm			1	1 Japin	毒		
				.•				
Example :	<b>Example 3:</b> 7000 cm = hm				<b>Example 4</b> : 8 daL = dL			
7000 CPT	1,091	1 hm	= 0.7  hm	8 det	10,1	10 dL	= 800 dL	
	100 cm	100 pr			1 det	1,1		
					7			

The factor label method can be used to solve virtually any problem including changes in units. It is especially useful in making complex conversions dealing with concentrations and derived units.

Convert the following.

1. 
$$35 \, \text{mL} = 0.35 \, \text{dL}$$

2. 
$$950 g = 0.95 kg$$

3. 
$$275 \, \text{mm} = \frac{27.5}{5} \, \text{cm}$$

6. 
$$4,500 \text{ mg} = 4.5 \text{ g}$$

7. 
$$25 \, \text{cm} = \frac{250}{} \, \text{mm}$$

8. 
$$0.005 \, \text{kg} = \frac{0.5}{0.5} \, \text{dag}$$

9. 
$$0.075 \, \text{m} = \frac{7.5}{1.5} \, \text{cm}$$

10. 
$$15g = 1.5 \times 10^{4} \text{ mg}$$