SCIENCE 8 OSMOSIS and GUMMY BEARS LAB

Name:		
Date:	Pd:	

Background:

Molecules are always moving. Remember that **DIFFUSION** is the movement of molecules from an area that is highly crowded to an area that is less crowded.

OSMOSIS is a form of DIFFUSION. OSMOSIS is the diffusion of <u>water molecules</u> through a selectively permeable membrane (a selectively permeable means some molecules can move through the membrane while others cannot.)

Cells use osmosis to move particles that it needs into and out of the cell.





Normal cell





Hypertonic
The concentration of solures
outside is higher
than it is inside
the cell.

Isotonic
The concentration of solutes
outside the cell is
equal to that
inside the cell.

Hypotonic
The concentration of solutes
outside is lower
than it is inside
the cell.

Very Hypotonic This cell has burst due to the large amount of water entering it.

Answer these questions BEFORE the lab:

	1) What is diffusion? (2 pts)	
	2) What is osmosis? (2pts)	
	3) What does selectively permeable mean? (2 pts)	
	4) Prediction: How will soaking Gummy Bear candies in distilled water affect the size of the (explain your prediction based on the background information) (6 pts)	ne candy?
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Materials:

TATCHCU TOTALS .	
1- 100 ml Beaker	2- Gummy bears
80 mL distilled water	Marker
Ruler	Paper Towel
80 mL Salt water	Calculator
Scale	2 dixie cups

PROCEDURE DAY 1:

- 1) Gather: 1 100mL beakers, 2 gummy bears, 2- dixie cups, a ruler, and a marker.
- 2) Label your cups with your group name.
- 3) Label one "distilled water" and the other "salt water."
- 4) Fill each cup with 80 mL of the designated water from the labeled beakers.
- 5) Measure both of your gummy bear's height, width and length in centimeters and record it in the data table below.
- 6) Using the digital scale, find both gummy bear's mass in grams and record it in the table below.
- 7) Draw all the details of your gummy bear in the data table below. Be sure to draw it the size it really is!

BEFORE SOAKING IN DISTILLED WATER	AFTER SOAKING IN DISTILLED WATER
Day 1 (4pts)	Day 2 (4pts)
HEIGHT:	HEIGHT:
WIDTH:	WIDTH:
LENGTH:	LENGTH:
MASS:	MASS:
DRAWING/DESCRIPTION	DRAWING/DESCRIPTION
DRAWING/DESCRIPTION	
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	AFTER SOAKING IN SALT WATER
BEFORE SOAKING IN SALT WATER	Day 2 (4 pts)
Day 1 (4 pts)	HEIGHT:
HEIGHT:	
WIDTH:	WIDTH:
LENGTH:	LENGTH:
MASS:	MASS:
DRAWING/DESCRIPTION	DRAWING/DESCRIPTION

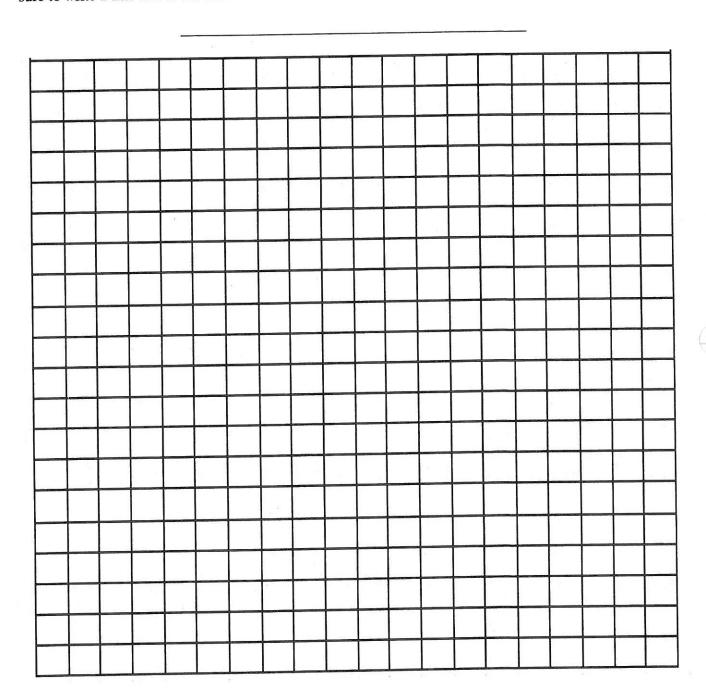
PROCEDURE DAY 2:

- 1) Find your group's beakers. Gently pour the water out into the designated receptacle.
- 2) Carefully transfer the gummy bear onto the paper towel. Be careful not to break the gummy bear.
- 3) Using your ruler and the scale, take the correct measurements to fill out the rest of your data table.
- 4) Calculate the percent change in the size of the candy using the formulas below:

Distilled Water - % Change (6pts)	
% CHANGE IN HEIGHT= (AFTER SOAKING HEIGHT- BEFORE SOAKING HEIGHT/ BEFORE SO	OAKING HEIGHT) x 100
(x 100 =	%
% CHANGE IN WIDTH= (AFTER SOAKING WIDTH- BEFORE SOAKING WIDTH/ BEF	AKING WIDTH) x 100
(x 100 =	%
% CHANGE IN LENGTH= (AFTER SOAKING LENGTH- BEFORE SOAKING LENGTH/ BEFORE S	SOAKING LENGTH) x 100
% CHANGE IN MASS= (AFTER SOAKING MASS- BEFORE SOAKING MASS/ BEFORE SOAKI	ING MASS) x 100
(x 100 =	%
Salt Water - % Change (6pts)	
% CHANGE IN HEIGHT= (AFTER SOAKING HEIGHT- BEFORE SOAKING HEIGHT/ BEFORE SO	OAKING HEIGHT) x 100
(x 100 =	%
% CHANGE IN WIDTH= (AFTER SOAKING WIDTH- BEFORE SOAKING WIDTH/ BEFORE SOA	KING WIDTH) x 100
(x 100 =	%
(x 100 = % CHANGE IN LENGTH= (AFTER SOAKING LENGTH- BEFORE SOAKING LENGTH/ BEFORE S	
% CHANGE IN LENGTH=	OAKING LENGTH) x 100

GRAPH: (10pts)

Graph the **percent changes** in height, width, length and mass for both gummy bears as bar graphs. Make sure to write a **title** and to **label both axes**.



PO

'OS	OST LAB QUESTIONS:	
1	. Describe what happened to your gummy bear after soaking it in distilled water over night: (4pts)	
2.	What changed the most, the height, width, length or mass of your distilled water gummy bear? (2pts)	
3.	Did the gummy bear soaked in distilled water have a hypotonic, hypertonic or isotonic reaction? (2pts)	
4.	Describe what happened to your gummy bear after soaking it in salt water over night: (4pts)	
5.	What changed the most, the height, width, length or mass of your salt water gummy bear? (2pts)	
6.	Did the gummy bear soaked in salt water have a hypotonic, hypertonic or isotonic reaction? (2pts)	

CONCLUSION: (15 pts)

Think about I balance with do not support	Write a <u>PARAGRAPH</u> which explains the results of this experiment using the concept of osmosis. Think about how much swelling can occur to the gummy bear for it to reach a point of equilibrium or palance with the solution you soaked the gummy bear in. Include how your results and data support or lo not support your prediction. Also, explain where you may have made mistakes in the lab and how you could make the lab more accurate.		
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