	Name:			
	Name: Blk:Date:			
Chemistry 12 Electrochemistry Lesson #5 Balancing Redox Equations using Half-Reactions				
IMPT: There are TWO way Reactions and using oxida	vs to Balance REDOX Equations: using Half- ation numbersyou learn both.			
This technique balances the redox equation by				
Example 1 : Balance Os + IO_3^- > OsO ₄ + I_2 (in acidic solution)				
Step 1:	politica.			
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Step 2:	The June 1			
Step 3:				
Step 4:				
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	of where the			
Example 2 . Balance $MnO_4^- + C_2O_4^{2-}> MnO_2 + CO_2$ (in basic solution)				
Step 1:				
	· Co.			
Step 2:				

Step 3:

Step 4:			
Step 5:			
Example 3: Balance C	10 ₂ > C10 ₃ - + C1	(in basic solution)	
Step 1:			
Step 2:			
Step 3:			
Step 4:			
Step 5:			
DISPROPORTIONATION	ON:		

Seatwork/Homework: do exercise 24 EVEN LETTERS ONLY!!!!

PLO's T2 and T3